PII: S0040-4039(96)02218-6

Towards Glycosyl Inositol Mimics: Stereoselective Synthesis of C-α-Glucosyl-inositol Derivatives via Diels-Alder Reactions

Francis Jacques, Hervé Dupeyroux, Jean-Pierre Joly and Yves Chapleur*

Laboratoire de Méthodologie et Synthèse Énantiospécifique de Biomolécules, associé au *CNRS* Université Henri Poincaré-Nancy I, BP 239, F-54506 Vandœuvre, France.

Abstract: The pentafluorophenyl ester of a C-glucopyranosidyl-linked olefin has proved to be an efficient chiral dienophile for asymmetric Diels-Alder reactions with E-1-acetoxy-1,3-butadiene to give C- α -glucosylcyclitol derivatives with excellent regioselectivity but fair diastereoselectivity. Copyright © 1996 Published by Elsevier Science Ltd

The biological role of glycosyl phosphatidyl inositols (GPI) as putative second messengers of insulin¹ and as key structure of protein anchorage system² is well recognised. Intracellular GPIs are hydrolysed by a specific phospholipase to glycosyl inositol phosphate which acts as a regulator of enzymes such as pyruvate dehydrogenase, adenylate cyclase and AMPc phosphodiesterase.³ On the other hand, the anchoring system based on GPI is used by parasites for the coating of their cells with glycoproteins which seem to protect the parasite from host defences³, and is likely used by the scrapie isoform of the prion protein.⁴ Hence it is clear that small molecules able to mimic the second messenger key-role of GPI are of interest for pharmacological studies.⁵ The same type of GPI analogues should also play a role as inhibitors of the biosynthesis of the GPI-anchoring system used by parasites such as those responsible for the African sleeping sickness.⁶

As seen from Scheme 1, the well conserved basic structure of GPI, I is a glucosamine- α - $(1\rightarrow 6)$ -inositol phosphate. Mimics of this molecule, such as II, should provide an entry for the elaboration of pharmaceutical tools and potential therapeutics. In connection with this problem, we have explored a route to II based on the total synthesis of the inositol ring via a Diels-Alder (D.-A.) reaction.^{7,8} Given the C-disaccharidic structure of II, the starting compound should be an allyl α -C-glycoside III derived from D-glucose via IV:

Scheme 1. Retrosynthesis of GPI mimetics from C-allyl glucoses

There are numerous examples in the literature of carbohydrates used as chiral auxilliary in D.-A. 9 or hetero D.-A. reactions. 10 However the involvment of C-glycosidic appendage in a [4+2] cycloaddition is less

documented.¹¹ In this letter, we report our first results on a D.-A. reaction between an activated double bond as dienophile and E-1-acetoxy-butadiene. Our synthesis started from the known C-allyl glycoside 1^{12} which was transformed into the ester 3 (E/Z=9:1) using standard procedures. These isomers were separated by chromatography but none of them showed interesting reactivity towards oxygenated dienes such as Danishefsky's diene, 1-methoxy or 1-acetoxy-butadiene even at 165° C in xylenes. This poor reactivity could be explained in terms of stereoelectronic effects. In the mean time Kan and Ohfune¹³ reported a tremendous improvement in D.-A. reactions by changing the methyl ester for the more active pentafluorophenyl ester. We were pleased to find that this was also true in our case. Thus the saponification of 3 to the corresponding acid 4 followed by dicyclohexylcarbodiimide-mediated coupling with pentafluorophenol¹⁴ gave 5 in 85% overall yield. The [4+2]-cycloaddition of the activated dienophile 5 with an excess of E-1-methoxy-1,3-butadiene was almost complete after 4 days at 120° C in toluene in a sealed vessel. A mixture of four partially separable cycloadducts was obtained in 74% yield and a roughly 1:1:1:1 ratio. In the same manner, but after heating 5 for ca. six days with a two-fold excess of E-1-acetoxy-1,3-butadiene, we successfully isolate besides 43% of unreacted starting material, two major pure fractions, 6a (17%) and 6b (17%), and two minor ones (in less than 8% of the products) which were not further investigated.

Scheme 2. Reagents and yields: i) O₃, MeOH, -50°C, 4 h; ii) PPh₃, -50 \rightarrow +20°C, 3 h; iii) Ph₃P=CHCO₂Me, ref. THF, 3 h, 84%, 3 steps; iv) NaOH 0.04 N / THF, 3 d, 91%; v) C₆F₅OH, DCCI, CH₂Cl₂, 0°C, 1.5 h, 93%; vi) 1-acetoxy-1,3-butadiene, toluene, 120°C, 6 d, 34%.

Both isolated cycloadducts **6a** and **6b** were diastereomerically pure on the basis of their 400 MHz 1 H NMR spectra measured in CDCl₃. The almost complete regioselectivity was obvious from the chemical shifts of H12 (δ 5.65, 5.7) and H13 (δ 3.15, 3.0) respectively. Since the pentafluorophenylester and the sugar moiety will be *trans* to each other due to the starting *E*-geometry of olefin **5**, the relative stereochemistry can be only *transtrans* or *cis-trans* for H12-H13 and H13-H8. The putative structures **6a** and **6b** are in agreement with the measured vicinal coupling constants $J_{8,13} = 12$ Hz (*trans*-diaxial orientation), and $J_{12,13}$ which was 4 Hz

(equatorial-axial cis orientation). These results are in full accord with some recent data obtained with the same diene.⁷ It was finally concluded that **6a** and **6b** differ only by the absolute configuration of the created asymmetric centres (vide infra). This result can be explained in terms of preferred endo transition states which led to a cis-relationship of the substituants at C12 and C13 in both cases. The 1:1 ratio found in the diastereoisomers **6a** and **6b** indicates the absence of facial selectivity in the studied D.-A. reaction. It is noteworthy to note that O-glycosidic dienes exhibited much higher facial selectivity.⁸

 $R = C_6 F_5$ $Su = 2,3,4,6-Tetra-O-benzyl-\alpha-D-glucopyranosyl$

Scheme 3. Two possible endo-attacks between dienophile 5 and E-1-acetoxy-1,3-butadiene

Crystals, suitable for X-ray crystallographic analysis of compounds 6, have not yet been obtained. Nevertheless, we explored further chemical transformations *en route* to the cyclitol ring. Complete reduction of 6a with LiAlH4 afforded the diol 7 in 90% yield. *Cis*-hydroxylation of the double bond using catalytic osmium tetroxide¹⁵ gave two isomers 8a $(J_{10,11} = 4.4 \text{ Hz})$ and 8b (not represented here) in a 89% overall yield and in a 4:1 ratio. The relative stereochemistry of the five chiral centres of the carbocyclic part in the major compound 8a was tentatively determined as all *cis*, given the small $J_{11,12} = 2.8 \text{ Hz}$ in favour of a *cis* orientation of H11 and H12. Hence, the major compound 8a should be depicted by one of the two following formulas:

Scheme 4. Reagents and yields: i) LiAlH₄, THF, $0\rightarrow20^{\circ}$ C, 1.5 h, 84%; ii) OsO₄, N-methyl morpholine oxide, acetone, water, 20°C, 36 h, 8a = 69%.

In conclusion, five of the six stereogenic centers present in the *myo*-inositol part of GPI anchors mimic 8a are installed (*epi* or *allo*-inositol stereo isomers) in a three-step sequence. Hence the use of an active ester, which dramatically improved the yield of the cycloaddition, is an important breakthrough that should be of general use in cycloaddition. Studies aimed at the improvement of the enantioselectivity (*ie* the facial selectivity) on the cycloaddition of sugar dienophiles using for instance chiral butadienes¹⁶ are currently in progress.

References

- 1. Alemany, S.; Mato, J. M.; Stålfors, P. Nature 1987, 330, 77-79.
- 2. Ferguson, M. A. J.; Homans, S. W.; Dwek, R. A.; Rademacher, T. W. Science 1988, 239, 753-759.
- 3. Dwek, R. A. Chem. Rev. 1996 96, 683-720.
- 4. Lehmann, S.; Harris, D. A. J. Biol. Chem. 1995, 270, 24589-24597.
- Reitz, A. B. Inositol Phosphates and Derivatives: Synthesis, Biochemistry and Therapeutic Potential;
 ACS Symposium Series 463, American Chemical Society: Washington D.C., 1991; Ley, S. V.; Yeung, L.
 L. Synlett 1992, 997-998; Reddy, K. K.; Falck, J. R.; Capdevila, J. Tetrahedron Lett. 1993, 34, 7869-7872; Jaramillo, C.; Chiara, J. L.; Martin-Lomas, M. J. Org. Chem. 1994, 59, 3135-3141; Hudlicky, T.;
 Thorpe, A. J. Synlett 1994, 899-901.
- Cottaz, S.; Brimacombe, J. S.; Ferguson, M. A. J. J. Chem. Soc., Perkin Trans. 1 1993, 2945-2951;
 Cottaz, S.; Brimacombe, J. S.; Ferguson, M. A. J. J. Chem. Soc., Perkin Trans. 1 1995, 1673-1678.
- 7. For a review on inositol phosphates, see: Potter, B. V. L.; Lampe, D. Angew. Chem., Int. Ed. Engl. 1995, 34, 1933-1972.
- 8. For a D.-A. approach to polyhydroxylated ring systems, see: a) Arjona, O.; Candilejo, A.; de Dios, A.; de la Pradilla, R. F.; Plumet, J. J. Org. Chem. 1992, 57, 6097-6099; b) Cottaz, S.; Brimacombe, J. S.; Ferguson, M. A. J. Carbohydr. Res. 1995, 270, 85-91; c) McClure, C.K.; Herzog, K. J.; Bruch, M. D. Tetrahedron Lett. 1996, 37, 2153-2156; d) Ferritto, R.; Vogel, P. Synlett, 1996, 3, 281-282, and references cited therein.
- For a review, see: Kunz, H.; Ruck, K. Angew. Chem. Int. Ed. Engl. 1993, 32, 336-358; for some examples of D.-A. reactions see: Kunz, H.; Müller, B.; Schanzenbach, D. Angew. Chem., Int. Ed. Engl. 1987, 26, 267-269; Gupta, R. C.; Raynor, C. M.; Stoodley, R. J.; Slawin, A. M. Z.; Williams, D. J. J. Chem. Soc., Perkin Trans. 1, 1988, 1773-1785; Gupta, R. C.; Larsen, D. S.; Stoodley, R. J.; Slawin, A. M. Z.; Williams, D. J. J. Chem. Soc., Perkin Trans. 1, 1989, 739-749; Larsen, D. S.; Stoodley, R. J. J. Chem. Soc., Perkin Trans. 1, 1990, 1339-1352; Larsen, D. S.; Stoodley, R. J. J. Chem. Soc., Perkin Trans. 1, 1990, 1841-1852; Charette, A. B.; Marcoux, J.-F.; Côté, B. Tetrahedron Lett. 1991, 32, 7215-7218; Nouguier, R.; Gras, J.-L.; Giraud, B.; Virgili, A. Tetrahedron Lett. 1991, 32, 5529-5530; Nouguier, R.; Gras, J.-L.; Giraud, B.; Virgili, A. Tetrahedron Lett. 1991, 32, 5529-5530; Nouguier, R.; Gras, J.-L.; Giraud, B.; Virgili, A. Tetrahedron, 1992, 48, 6245-6252; Beagley, B.; Curtis, A. D. M., Pritchard, R. G.; Stoodley, R. J. J. Chem. Soc., Perkin Trans. 1, 1992, 1981-1991; Larsen, D. S.; Trotter, N. S.; Stoodley, R. J. Tetrahedron Lett. 1993, 34, 8151-8154; Aspinall, I. H.; Cowley, P. M.; Stoodley, R. J. Tetrahedron Lett. 1994, 35, 3397-3400; Lubineau, A.; Bienaymé, H.; Queneau, Y.; Schermann, M.-C. New J. Chem. 1994, 18, 279-285; Shing, T. K. M.; Chow, H.-F.; Chung, I. H. F. Tetrahedron Lett. 1996, 37, 3713-3716.
- For some examples of hetero D.-A. see: a) David, S.; Lubineau, A.; Thieffry, A. Tetrahedron, 1978, 34, 299-307; b) Pfrengle, W.; Kunz, H. J. Org. Chem. 1989, 54, 4261-4263.
- Danishefsky, S. J.; Maring, C. J.; Barbachyn, M. R.; Segmuller, B. E. J. Org. Chem. 1984, 49, 4564-4565;
 Danishefsky, S. J.; Peason, W. H.; Harvey, D. F.; Maring, C. J.; Springer, J. P. J. Am. Chem. Soc. 1985, 107, 1256-1268.
- 12. Hosomi, A.; Sakata, Y.; Sakurai, H. Carbohydr. Res. 1987, 171, 223-232.
- 13. Kan, T.; Ohfune, Y. Tetrahedron Lett. 1995, 36, 943-946.
- 14. Kisfaludy, L.; Löw, M.; Nyéki, O.; Szirtes, T.; Schön, I. Liebigs Ann. Chem. 1973, 1421-1429.
- 15. VanRheenen, V.; Kelly, R. C.; Cha, D. Y. Tetrahedron Lett. 1976, 23, 1973-1976.
- 16. Fisher, M. J.; Hehre, W. J.; Kahn, S. D.; Overmann, L. E. J. Am. Chem. Soc. 1988, 110, 4625-4633.